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Key indicators

Single-crystal X-ray study
 $T = 120$ K
Mean $\sigma(\text{C}-\text{C}) = 0.008$ Å
 R factor = 0.101
 wR factor = 0.192
Data-to-parameter ratio = 15.9

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

4-Aminomethyl-phenylamino-bis-(3,4-dichloro-5-phenylcarbamoyl-1*H*-pyrrole-2-carboxamide) tetrabutylammonium salt

The crystal structure of the tetrabutylammonium salt of doubly deprotonated 4-aminomethyl(phenylamino)bis-(3,4-dichloro-5-phenylcarbamoyl-1*H*-pyrrole-2-carboxamide), $2\text{C}_{16}\text{H}_{36}\text{N}^+ \cdot \text{C}_{32}\text{H}_{22}\text{Cl}_4\text{N}_6\text{O}_4^{2-}$, has been elucidated. The anion lies on an inversion centre and adopts a twisted S shape.

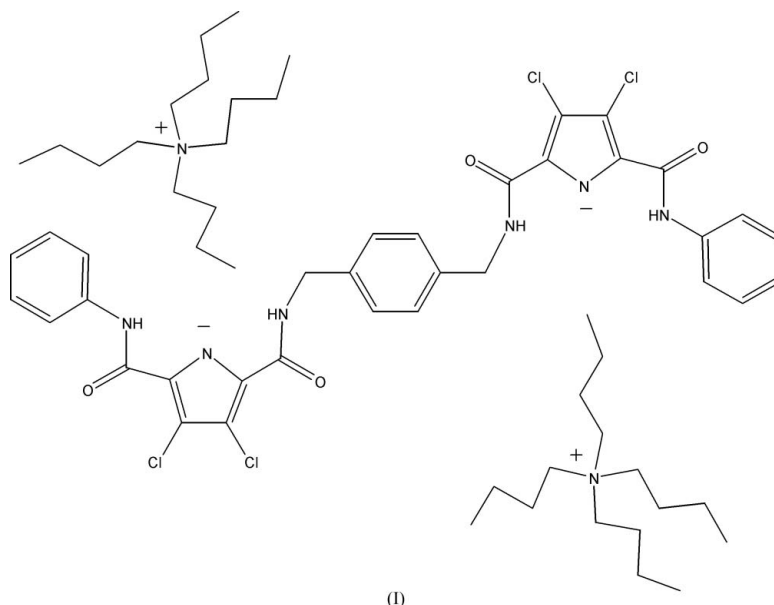
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Comment

4-Aminomethyl(phenylamino)bis-(3,4-dichloro-5-phenylcarbamoyl-1*H*-pyrrole-2-carboxamide) crystallizes as the tetrabutylammonium salt, (I), from an acetonitrile solution of the compound in the presence of excess tetrabutylammonium fluoride. The anion adopts a twisted S shape around a centre of inversion. The pyrrole and terminal benzene ring pairs are coplanar and the angle between the central and terminal benzene ring is $73.01(5)^\circ$.



Experimental

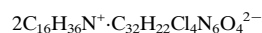
p-Xylenediamine (68 mg, 0.5 mmol, 1 equiv.) was added to a solution of 3,4-dichloro-5-phenylcarbamoyl-1*H*-pyrrole-2-carboxylic acid (300 mg, 1 mmol, 2 equiv.) in DMF (30 ml) under a nitrogen atmosphere. Triethylamine (104 mg, 1 mmol, 2 equiv.), benzotriazol-1-yloxy)tripyrrolidinophosphonium hexafluorophosphate (572 mg, 1.1 mmol, 2.2 equiv.) and 5 mg (0.04 mmol, 0.04 equiv.) of *N*-hydroxybenzotriazole were added and the reaction was stirred for 72 h. The solvent was then removed and water (50 ml) was added. The product was extracted with dichloromethane (DCM, 3 × 50 ml). The organic phase was collected and the solvent was removed. The product was washed with diethyl ether (75 ml) and a small quantity of 10% MeOH

in DCM (v/v). The product was obtained as a white solid (132 mg, 0.19 mmol, 38%).

M.p. 590 K (decomp.). ^1H NMR 300 MHz in $\text{DMSO-}d_6$ δ (p.p.m.): 4.50 (*d*, $J = 5.4$, 4H, CH_2), 7.00–7.70 (*m*, 14H, ArH), 8.50 (*t*, 2H, $J = 5.4$, central-CONH), 10.04 (*s*, 2H, outer-CONH), 12.79 (*s*, 2H, NH-pyrrole). ^{13}C NMR 75 MHz in $\text{DMSO-}d_6$ δ (p.p.m.): 42.3, 112.0, 113.6, 119.8, 122.9, 123.1, 123.9, 127.4, 128.7, 137.6, 138.3, 156.5, 158.1. TOF LD^+ mass spectrum: m/z (%): 472 (100) $[\text{C}_{23}\text{H}_{22}\text{Cl}_2\text{N}_4\text{O}_3]^+$. Elemental analysis: Calc. for $\text{C}_{32}\text{H}_{24}\text{Cl}_4\text{N}_6\text{O}_4 \cdot \text{H}_2\text{O}$: C 53.65, H 3.66, N 11.73%; found: C 53.28, H 3.73, N 12.03%.

Crystals of the title compound were obtained by slow evaporation of an acetonitrile solution in the presence of excess tetrabutylammonium fluoride.

Crystal data



$M_r = 1181.27$

Triclinic, $P\bar{1}$

$a = 9.524$ (2) Å

$b = 10.363$ (3) Å

$c = 17.056$ (4) Å

$\alpha = 78.293$ (11)°

$\beta = 88.733$ (14)°

$\gamma = 80.136$ (14)°

$V = 1623.9$ (7) Å³

$Z = 1$

$D_x = 1.208$ Mg m⁻³

Mo $K\alpha$ radiation

Cell parameters from 58835

reflections

$\theta = 2.9$ – 27.5°

$\mu = 0.23$ mm⁻¹

$T = 120$ (2) K

Block, colourless

$0.20 \times 0.10 \times 0.07$ mm

Data collection

Bruker–Nonius KappaCCD area-detector diffractometer

φ and ω scans

Absorption correction: multi-scan (SORTAV; Blessing, 1997)

$T_{\min} = 0.732$, $T_{\max} = 0.984$

16494 measured reflections

5757 independent reflections

2777 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.111$

$\theta_{\max} = 26.1^\circ$

$h = -11 \rightarrow 11$

$k = -12 \rightarrow 12$

$l = -20 \rightarrow 20$

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.101$

$wR(F^2) = 0.192$

$S = 1.04$

5757 reflections

361 parameters

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0424P)^2 + 2.4189P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta\sigma)_{\max} = 0.019$$

$$\Delta\rho_{\max} = 0.25 \text{ e } \text{Å}^{-3}$$

$$\Delta\rho_{\min} = -0.24 \text{ e } \text{Å}^{-3}$$

H atoms were identified in a difference map and then placed in calculated positions (N–H 0.88, aromatic C–H 0.95, methylene C–H 0.99, methyl C–H 0.98) and refined using a riding model [$U_{\text{iso}}(\text{H}) = 1.2$ or 1.5 times $U_{\text{eq}}(\text{C}, \text{N})$].

Data collection: *DENZO* (Otwinowski & Minor, 1997) and *COLLECT* (Hoof, 1998); cell refinement: *DENZO* and *COLLECT*;

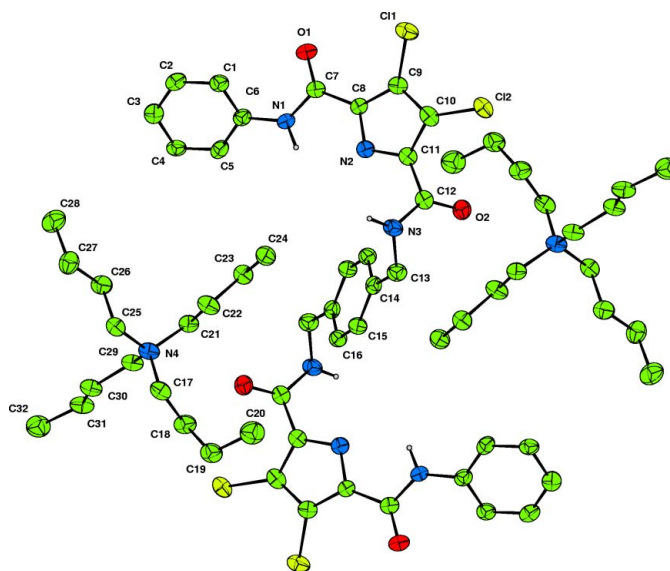


Figure 1

Structure of the title compound, with displacement ellipsoids drawn at the 50% probability level and non-acidic H atoms omitted for clarity.

data reduction: *DENZO* and *COLLECT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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